

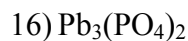
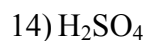
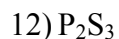
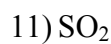
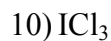
Name _____

AP Chemistry

Inorganic Chemistry Nomenclature Practice

Part I

Name each of the following compounds.



Part II

Write the formula for the following compound.

1) Zinc chloride

2) Tin (IV) fluoride

3) Calcium nitride

4) Aluminum sulfide

5) Mercury (I) selenide

- 6) Silver iodide
- 7) Ammonium hydrogen phosphate
- 8) Mercury (I) sulfide
- 9) Silicon dioxide
- 10) Sodium sulfite
- 11) Aluminum hydrogen sulfate
- 12) Nitrogen trichloride
- 13) Hydrobromic acid
- 14) Bromous acid
- 15) Perbromic acid
- 16) Potassium hydrogen sulfide
- 17) Calcium iodide
- 18) Cesium perchlorate